

Synthesis, Radiosynthesis, and Biological Evaluation of Fluorine-18-Labeled 2 β -Carbo(fluoroalkoxy)-3 β -(3'-((Z)-2-haloethenyl)phenyl)nortropans: Candidate Radioligands for In Vivo Imaging of the Serotonin Transporter with Positron Emission Tomography

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The *meta*-vinylhalide fluoroalkyl ester nortropans **1–4** were synthesized as ligands of the serotonin transporter (SERT) for use as positron emission tomography (PET) imaging agents. In vitro competition binding assays demonstrated that **1–4** have a high affinity for the SERT (K_i values = 0.3–0.4 nM) and are selective for the SERT over the dopamine and norepinephrine transporters (DAT and NET). MicroPET imaging in anesthetized cynomolgus monkeys with [¹⁸F]**1**–[¹⁸F]**4** demonstrated that all four tracers behave similarly with peak uptake in the SERT-rich brain regions achieved after 45–55 min, followed by a steady washout. An awake monkey study was performed with [¹⁸F]**1**, which demonstrated that the uptake of [¹⁸F]**1** was not influenced by anesthesia. Chase studies with the SERT ligand **15** displaced [¹⁸F]**1**–[¹⁸F]**4**, but chase studies with the DAT ligand **16** did not displace [¹⁸F]**1**–[¹⁸F]**4** thus indicating that the tracers were binding specifically to the SERT.

Introduction

The human serotonin transporter (SERT)^a is a 630 amino acid transmembrane protein located in the brain, peripheral organs, and blood platelets.^{3–6} Within the central nervous system (CNS), the SERT is localized on the presynaptic terminals of serotonergic neurons and functions to terminate neurotransmission by removing serotonin from the synapse. Serotonergic neurons originate primarily in the median and dorsal raphe nuclei of the brainstem and innervate discrete areas that include the hypothalamus, thalamus, striatum, and cerebral cortex.^{4,7–10} Thus, the SERT can serve as a specific marker for serotonergic neuronal anatomy and integrity. Dysregulation of serotonin neurotransmission has been implicated in the pathophysiology of major depression and a reduction in SERT density has been observed postmortem in the tissues of depressed patients and suicide victims.^{7,11–14} Therefore, the SERT is the target of the selective serotonin reuptake inhibitor (SSRI) class of antidepressants. The ability to image CNS SERT in vivo using positron emission tomography (PET)^{15–17} may provide insight into the pathophysiology of depression and suicide by enabling the SERT density of specific brain regions to be measured, thereby indicating which regions of the brain have SERT density altered by the disease, as well as allow for improved diagnostic techniques and monitoring of antidepressant therapy.^{18–22} Furthermore, the availability of SERT PET tracers may aid in

the development of new SERT therapeutics by allowing for occupancy measurements of the therapeutic.^{23–28}

This need for a SERT PET tracer has led to extensive research into the development of new tracers for this target¹⁷ with the majority of these new compounds belonging to the diarylsulfide^{24,29–43} or nortropane^{44–46} classes. Many of the initial SERT PET tracers are ¹¹C-labeled compounds and therefore are limited to use in the location where they are prepared due to the short half-life of ¹¹C ($t_{1/2}$ = 20.4 min). Tracers labeled with ¹⁸F are desirable because the longer half-life of ¹⁸F ($t_{1/2}$ = 109.8 min) allows for longer radiosynthesis times and transport of ¹⁸F-labeled tracers to sites away from the production facility which thus enables PET imaging centers without onsite cyclotrons to utilize these tracers. Furthermore, ¹⁸F positrons have a lower maximum energy than ¹¹C positrons (0.64 vs 0.97 MeV)¹⁷ and therefore deposit less energy into tissue, and they also have a shorter linear range^{47,48} that allows for higher spatial resolution. These properties of ¹⁸F are fortuitous because of the valuable role that ¹⁹F plays in medicinal chemistry^{49–51} and numerous methods have now been developed to incorporate ¹⁸F or ¹⁹F into molecules.^{52–54}

The goal of developing a viable SERT PET tracer is an important, but not easy, task as evidenced by the numerous compounds that have been and continue to be reported. Several criteria need to be met for a candidate molecule to become a useful tracer. The desirable properties for a candidate SERT PET tracer include (exceptions may exist) (1) high binding affinity⁵⁵ (K_i = ~0.1–0.5 nM) for the SERT with high selectivity (>50:1) over the norepinephrine transporter (NET) and dopamine transporter (DAT); (2) moderate lipophilicity (log P = ~1–3)^{56–58} for good initial brain entry and low nonspecific binding; (3) high uptake ratios versus cerebellum⁵⁹ (≥ 1.7) in SERT-rich brain regions, such as thalamus, hypothalamus, and raphe, with uptake ratios ≥ 1.3 in low density regions such as anterior cingulate to enable delineation of specific SERT binding; (4) specific binding to brain SERT reaching peak equilibrium at ≤ 60 min, followed by washout for both ¹¹C-

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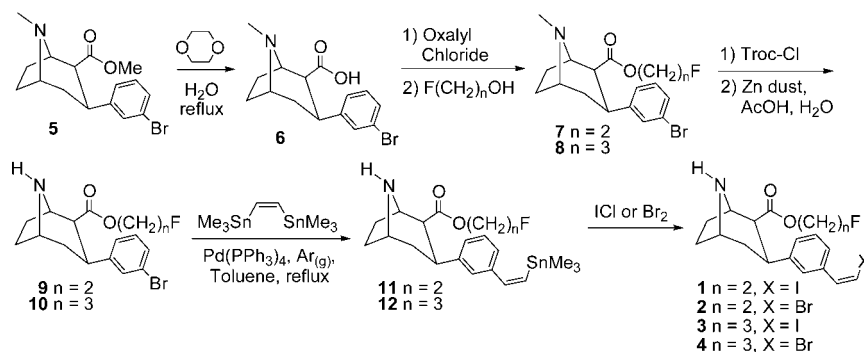
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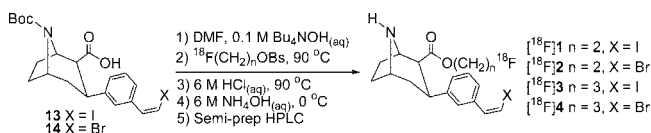
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^a Abbreviations: PET, positron emission tomography; SUV, standardized uptake value;^{1,2} SPECT, single-photon emission computed tomography; TAC, time-activity curve; HRRT, high resolution research tomograph; EOB, end-of-bombardment; CNS, central nervous system; SSRI, selective serotonin reuptake inhibitor; SERT, serotonin transporter; DAT, dopamine transporter; NET, norepinephrine transporter.

Scheme 1



Scheme 2



and ^{18}F -tracers to allow quantification of SERT occupancy; (5) lack of radiolabeled metabolites generated in the brain; and (6) lack of lipophilic radiolabeled metabolites in peripheral organs that may enter brain and bind specifically or nonspecifically to SERT-rich regions and cerebellum.

As part of an ongoing research project in our laboratories to develop SERT-selective tropane and nortropine PET and single-photon emission computed tomography (SPECT) imaging agents labeled with ^{11}C or ^{18}F (PET) or ^{123}I (SPECT) for human diagnostic applications we have been exploring the effect of placing a vinyl halide^{60–66} or furyl substituent⁶⁷ on the 3β-phenyl ring. We report here the synthesis and biological evaluation of 2β-carbo(2-fluoroethoxy)-3β-(3'-((Z)-2-iodoethenyl)phenyl)nortropine (**1**, FEMZIEN),⁶⁸ 2β-carbo(2-fluoroethoxy)-3β-(3'-((Z)-2-bromoethenyl)phenyl)nortropine (**2**, FEMZBrENT),⁶⁹ 2β-carbo(3-fluoropropoxy)-3β-(3'-((Z)-2-iodoethenyl)phenyl)nortropine (**3**, FPMZIEN),⁷⁰ and 2β-carbo(3-fluoropropoxy)-3β-(3'-((Z)-2-bromoethenyl)phenyl)nortropine (**4**, FPMZBrENT),⁷⁰ along with the radiosynthesis and microPET imaging of [^{18}F]**1**–[^{18}F]**4** in anesthetized nonhuman primates and the PET imaging of [^{18}F]**1** in an awake nonhuman primate.

Chemistry

The synthesis of target compounds **1**–**4** is shown in Scheme 1. 2β-Carbomethoxy-3β-(3'-bromophenyl)tropine (**5**)⁶³ was hydrolyzed in refluxing 1,4-dioxane/ H_2O ⁷¹ to give the acid **6** which was converted to the acid chloride and then esterified to afford the 2-fluoroethyl ester **7** or the 3-fluoropropyl ester **8**. *N*-demethylation⁷² afforded the nortropes **9** and **10**, which were coupled to (Z)-1,2-bis(trimethylstannyl)ethene^{73,74} to give the vinyl–tin nortropes **11** and **12** in varying cis/trans ratios with the major product being the cis isomer (~3:1). Halodestannylation then afforded the vinyl iodides **1** and **3** and the vinyl bromides **2** and **4** (after separation of the cis/trans isomers by semipreparative HPLC).

Radiochemistry

The radiolabeling procedure is depicted in Scheme 2. *N*-Boc acid **13** or **14**⁶³ was dissolved in DMF, deprotonated with 0.1 M $\text{Bu}_4\text{OH}_{(\text{aq})}$, added to [^{18}F]fluoroalkylbrosylate, and heated. The *N*-Boc group was cleaved under acidic conditions, the solution was neutralized, and the mixture was purified by semipreparative HPLC. The desired HPLC fractions were

Table 1. Octanol/Water Partition Coefficients

compound	$\log P_{7,4}^a$	<i>n</i>
[^{18}F] 1	1.69 ± 0.01	7
[^{18}F] 2	1.40 ± 0.03	4
[^{18}F] 3	1.64 ± 0.14	12
[^{18}F] 4	1.87 ± 0.03	8

^a Average value of *n* determinations \pm the standard deviation.

combined and the product was isolated by solid phase extraction according to a previously reported procedure.⁷⁵ We were able to obtain higher decay-corrected radiochemical yields with [^{18}F]FETOBs than with [^{18}F]FPROBs in our radiosyntheses (see Experimental Section). This is presumably because the carbon atom bonded to the brosylate leaving group bears a larger partial positive charge in [^{18}F]FETOBs than [^{18}F]FPROBs due to the electron-withdrawing properties of the fluorine atom which is two bonds away in [^{18}F]FETOBs but three bonds away in [^{18}F]FPROBs. The octanol/water partition coefficients^{56–58} of [^{18}F]**1**–[^{18}F]**4** were measured according to a known procedure^{76,77} and are shown in Table 1. These values are all in the range⁵⁶ that will allow for diffusion of the tracer across the blood-brain-barrier.

In Vitro Competition Binding Assays

Vinylhalide nortropes **1**–**4** and *trans*-**1** were screened for binding to human monoamine transporters using in vitro competition binding assays with transfected SERT, DAT, or NET according to our previously reported procedure.^{61,78} The binding affinities for each transporter were determined using [^3H](*R,S*)-citalopram·HBr^{78,79} ([^3H]**15**) (SERT), [^{125}I]RTI-55⁸⁰ (DAT), or [^3H]Nisoxetine⁸¹ (NET). The data in Table 2 indicate that **1**–**4** all have a high affinity for the SERT and are selective for the SERT over the DAT and NET.⁵⁵ *trans*-**1** has a reduced affinity at all three transporters as expected based on previous reports.^{60,74}

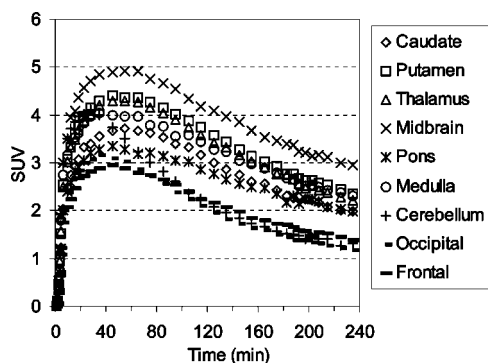
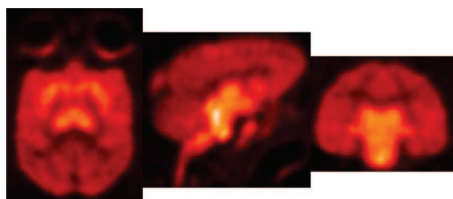
In Vivo Nonhuman Primate MicroPET Imaging

The in vivo regional brain uptake of [^{18}F]**1**–[^{18}F]**4** was determined in anesthetized cynomolgus monkeys (a total of 4 different cynomolgus monkeys) using a Concorde microPET P4 according to our previously reported procedure.⁶¹ Baseline studies were initially performed to determine the extent of uptake of [^{18}F]**1**–[^{18}F]**4** in the SERT-rich regions of the brain. The baseline time-activity curves (TACs) for [^{18}F]**1** are shown in Figure 1. Compound [^{18}F]**1** enters the brain rapidly and achieves peak uptake in the SERT-rich brain regions after ~45 min (Table S1, Supporting Information) with the highest uptake observed in the midbrain followed by the putamen, thalamus, medulla, and caudate. This rank order of uptake is similar to that observed with previously reported ^{11}C -labeled diarylsul-

Table 2. Results of In Vitro Competition Binding Assays with Transfected Human Monoamine Transporters

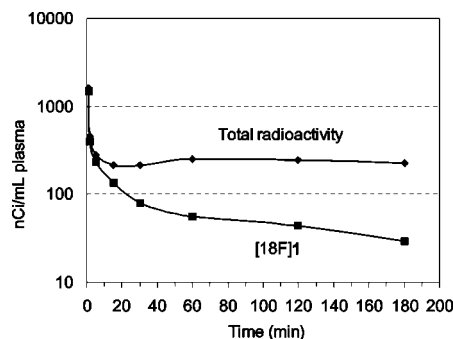
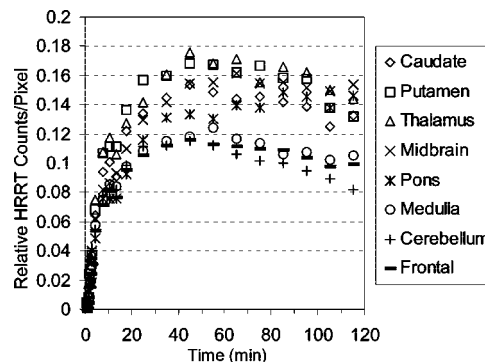
compd	K_i (nM) ^a			SERT selectivity	
	hSERT	hDAT	hNET	DAT/SERT	NET/SERT
1	0.43 ± 0.01 ^b	87.6 ± 10.2 ^b	109.9 ± 6.3 ^c	~204	~256
2	0.33 ± 0.10 ^b	35.5 ± 5.9 ^b	101.3 ± 2.5 ^b	~108	~307
3	0.26 ± 0.03 ^b	179.7 ± 17.8 ^b	67.5 ± 5.6 ^b	~691	~260
4	0.33 ± 0.03 ^b	118.7 ± 13.8 ^b	30.2 ± 0.4 ^b	~360	~92
trans-1	15.2 ± 3.9 ^b	274.0 ± 9.3 ^b	1180 ± 426 ^b	~18	~78

^a Average value of *n* determinations ± SEM (each determination performed in triplicate). ^b *n* = 2. ^c *n* = 3.

**Figure 1.** MicroPET baseline TACs obtained by injection of [¹⁸F]1 into an anesthetized cynomolgus monkey.**Figure 2.** MicroPET images (summed 0–235 min) obtained by injection of [¹⁸F]1 into an anesthetized cynomolgus monkey.

fides^{30,31,33,34,37} and is consistent with the known distribution of SERT in the brain.^{34,59,82,83} Lesser uptake is observed in the pons and the occipital and frontal cortices. The uptake in the midbrain, putamen, thalamus, medulla, and caudate remains nearly constant for ~20 min (~45–65 min postinjection) and then begins to steadily wash out. In the cerebellum, a region with low SERT density,⁵⁹ peak uptake is achieved after ~27 min, followed by a rapid washout of radioactivity down to uptake levels similar to that observed in the occipital and frontal cortices. Between 65–175 min postinjection the washout from the SERT-rich brain regions remains nearly parallel with the washout from the cerebellum indicating that a quasi-equilibrium (a condition where the ratio of radioactivity uptake in the region of interest to reference region stays relatively constant)³⁷ has been established. The microPET images from this baseline study are shown in Figure 2.

Metabolite analysis of [¹⁸F]1 (Figure 3) was performed with arterial plasma samples and determined by an HPLC method with radioactivity detection as previously described.⁶¹ The initial arterial plasma sample was taken at 1 min postinjection of [¹⁸F]1 and consisted of 93% of the total plasma radioactivity as unmetabolized [¹⁸F]1. The percent of total plasma radioactivity then decreased over time during the course of the study to 13% unmetabolized [¹⁸F]1 after 180 min. The radioactive metabolite eluted immediately after the void volume during HPLC analysis and is believed to be [¹⁸F]fluoroethanol (or one of its metabolic oxidation products [¹⁸F]fluoroacetaldehyde or [¹⁸F]fluoroacetic acid) that would result from hydrolysis of the [¹⁸F]fluoroethyl ester.⁸⁴ The percentage of protein-bound [¹⁸F]1 in plasma at

**Figure 3.** Metabolite analysis of [¹⁸F]1 in an anesthetized cynomolgus monkey.**Figure 4.** HRRT baseline TACs obtained by injection of [¹⁸F]1 into an awake rhesus monkey.

each time point is shown in Table S15, Supporting Information, and was in the range of 5.6–8.5% during the course of the study.

It has been previously shown that anesthesia during PET imaging can interfere with radioligand binding.^{85–88} We therefore performed a baseline study in an awake rhesus monkey with [¹⁸F]1 using a Siemens/CTI High Resolution Research Tomograph (HRRT) to determine if there was a difference in the behavior of [¹⁸F]1 in an awake versus an anesthetized state. The TACs are shown in Figure 4 and the PET images are shown in Figure 5. High uptake is observed in the thalamus, putamen, midbrain, and caudate with peak uptake achieved after ~45 min (Table S2, Supporting Information). The behavior of [¹⁸F]1 in an awake monkey is very similar to its behavior in an anesthetized monkey thus demonstrating that the imaging properties of [¹⁸F]1 are not affected by anesthesia. We were not able to follow the washout of [¹⁸F]1 in the awake study for the same amount of time as with the anesthetized study because of the difficulty of keeping an awake monkey motionless for an extended period of time.

The TACs for the baseline studies with [¹⁸F]2–[¹⁸F]4 are shown in Figures 6–8, respectively. Compound [¹⁸F]2 reaches peak uptake after 55–65 min in the SERT-rich brain regions followed by a steady washout (Table S3, Supporting Information). Peak uptake in the cerebellum is achieved after ~30 min,

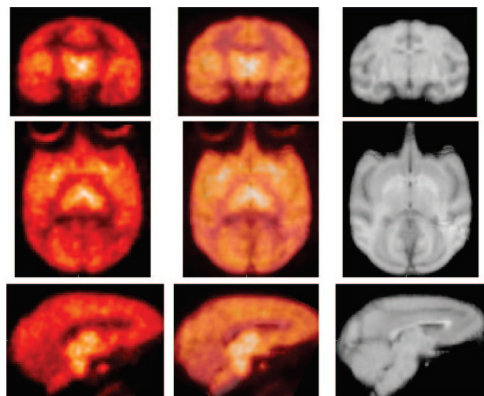


Figure 5. HRRT PET images (left, summed 60–120 min) obtained by injection of [^{18}F]1 into an awake rhesus monkey. Composite MRI's of several rhesus monkeys (right) and overlaid images (center).

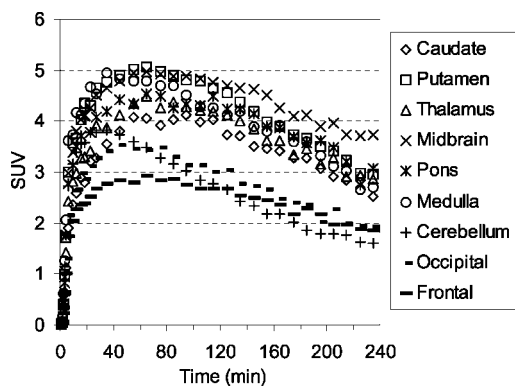


Figure 6. MicroPET baseline TACs obtained by injection of [^{18}F]2 into an anesthetized cynomolgus monkey.

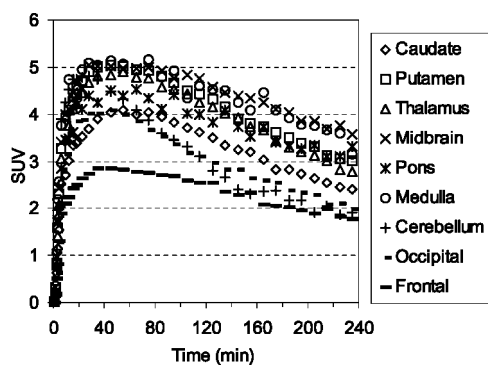


Figure 7. MicroPET baseline TACs obtained by injection of [^{18}F]3 into an anesthetized cynomolgus monkey.

followed by a rapid washout to a level of uptake slightly less than that observed in the occipital and frontal cortices. Peak uptake in the SERT-rich brain regions for [^{18}F]3 was achieved after 45–55 min (Table S4, Supporting Information) and for [^{18}F]4 peak uptake was achieved after 35–45 min (Table S5, Supporting Information) with both compounds showing a steady washout. The ratio of uptake in the SERT-rich brain regions vs cerebellum uptake for compounds [^{18}F]1–[^{18}F]4 at 115 and 215 min postinjection is shown in Table 3 (the complete data is shown in Tables S6–S10, Supporting Information, along with graphs of the uptake ratios versus time in Figures S1–S5, Supporting Information). The ratios for each tracer are very similar at these two time points thus indicating that these tracers all behave very similarly in vivo. The differences between the data in Table 3 are believed to be the result, at least partially,

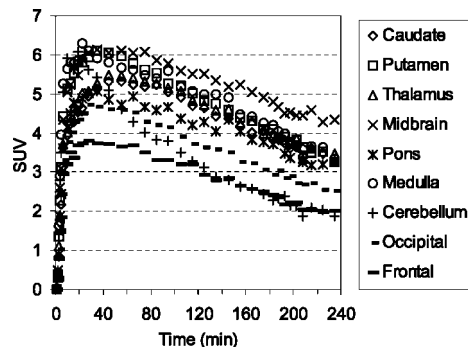


Figure 8. MicroPET baseline TACs obtained by injection of [^{18}F]4 into an anesthetized cynomolgus monkey.

Table 3. Comparison of Uptake of [^{18}F]1–[^{18}F]4 in SERT-Rich Brain Regions to Cerebellum Uptake at 115 and 215 min Post-Injection

brain region	115 min					215 min			
	[^{18}F]1 ^a	[^{18}F]1	[^{18}F]2	[^{18}F]3	[^{18}F]4	[^{18}F]1	[^{18}F]2	[^{18}F]3	[^{18}F]4
caudate	1.6	1.4	1.5	1.2	1.4	1.5	1.6	1.2	1.7
putamen	1.6	1.7	1.7	1.4	1.5	1.8	1.8	1.5	1.7
thalamus	1.8	1.6	1.5	1.4	1.5	1.7	1.6	1.5	1.7
midbrain	1.9	1.9	1.7	1.5	1.7	2.2	2.1	1.8	2.2
pons	1.8	1.3	1.6	1.3	1.3	1.5	1.8	1.5	1.5
medulla	1.3	1.5	1.5	1.5	1.5	1.7	1.7	1.8	1.8
occipital	N/A	0.9	1.1	1.0	1.1	0.9	1.1	1.0	1.3
frontal	1.2	1.0	1.0	0.8	1.0	1.1	1.1	1.0	1.0

^a HRRT awake rhesus study.

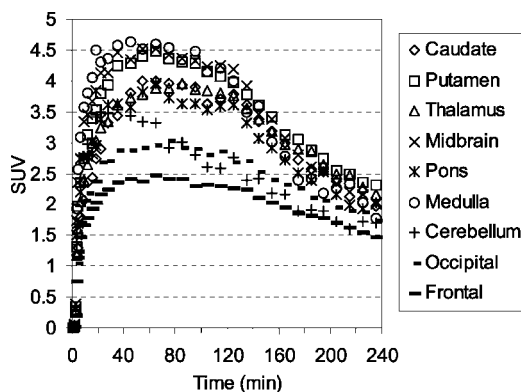


Figure 9. MicroPET TACs showing the results of injection of **15** (1.5 mg/kg) into an anesthetized cynomolgus monkey at 120 min postinjection of [^{18}F]1.

of individual differences between the monkeys studied rather than differences between the performance of each tracer. The uptake ratios for [^{18}F]1 after 115 min in awake and anesthetized monkeys are also very similar which provides further evidence that isoflurane anesthesia does not influence the in vivo behavior of [^{18}F]1.

To demonstrate that the observed uptake in the baseline studies with [^{18}F]1–[^{18}F]4 is the result of preferential binding to the SERT and not the DAT, chase studies were performed with the SERT ligand **15**, and the DAT ligands (\pm)-methylphenidate $\cdot\text{HCl}$ ⁸⁹ (**16**) and RTI-113 $\cdot\text{HCl}$ ^{71,90,91} (**17**). Figures 9–12 show the TACs for the chase studies with **15** at 120 min postinjection of the tracer for compounds [^{18}F]1–[^{18}F]4, respectively. For all four chase studies the amount of radioactivity in the SERT-rich brain regions decreases to nearly the level of the cortices and cerebellum thus indicating that the observed uptake is the result of binding to the SERT. These results also indicate that [^{18}F]1–[^{18}F]4 can be used for occupancy studies of SERT-selective therapeutics.²⁶ Graphs of the uptake ratio

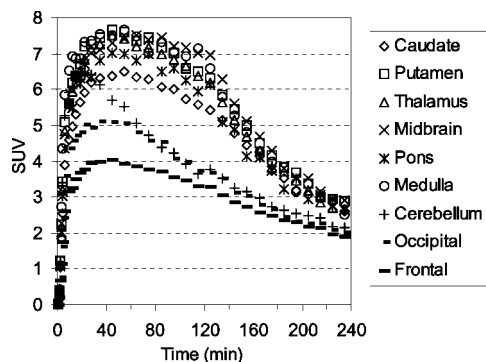


Figure 10. MicroPET TACs showing the results of injection of **15** (1.5 mg/kg) into an anesthetized cynomolgus monkey at 120 min postinjection of [^{18}F]**2**.

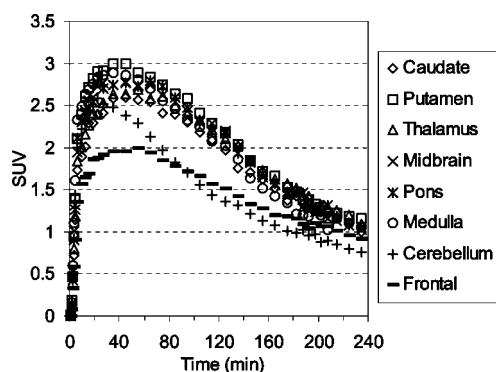


Figure 11. MicroPET TACs showing the results of injection of **15** (1.5 mg/kg) into an anesthetized cynomolgus monkey at 120 min postinjection of [^{18}F]**3**.

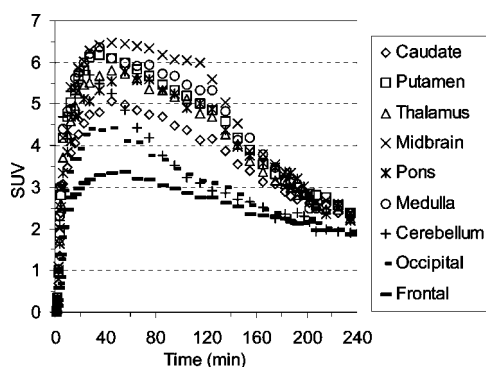


Figure 12. MicroPET TACs showing the results of injection of **15** (1.5 mg/kg) into an anesthetized cynomolgus monkey at 120 min postinjection of [^{18}F]**4**.

versus time for these chase studies are shown in Figures S6–S9, Supporting Information. Figure S10, Supporting Information, shows the results of chasing [^{18}F]**1** with the DAT ligand **16**, and Figure S11, Supporting Information, shows the results of chasing [^{18}F]**1** with the DAT ligand **17**. We chose to perform chase studies with both **16** and **17** because we wanted to chase [^{18}F]**1** with a tropane-based DAT ligand (**17**) and a nontropane-based DAT ligand (**16**). In neither case did the amount of radioactivity uptake decrease after administration of the chase compound (other than normal washout) thus indicating that [^{18}F]**1** is not bound to the DAT. Chase studies using **16** were also performed with [^{18}F]**2**–[^{18}F]**4** (Figures S12–S14, Supporting Information, respectively) and, similarly to what was observed with [^{18}F]**1**, the radioactivity was not displaced. We did not perform any chase studies with the NET ligand **18**.^{92,93}

Table 4. Comparison of the Times of Peak Uptake between [^{18}F]**1** and [^{18}F]**19** for Anesthetized and Awake Monkey Studies^a

brain region	time (min)			
	anesthetized		awake	
	[^{18}F] 1	[^{18}F] 19	[^{18}F] 1	[^{18}F] 19
caudate	55	105	45	85
putamen	45	95	45	85
thalamus	45	85	45	>75
midbrain	45	95	45	>75
pons	45	115	85	>75
medulla	35	85	35	85
cerebellum	28	45	45	35

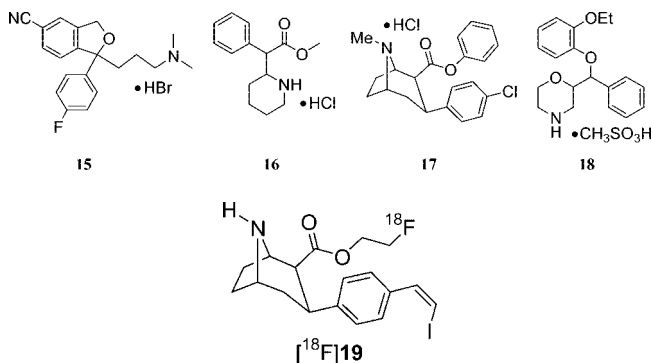
^a See Tables S1, S2, S11, and S13, Supporting Information, for the complete data.

Table 5. Comparison of the Uptake Ratios between [^{18}F]**1** and [^{18}F]**19** for Anesthetized and Awake Monkey Studies at Selected Time Points^a

brain region	anesthetized						awake	
	55 min		115 min		225 min		95 min	
	[^{18}F] 1	[^{18}F] 19	[^{18}F] 1	[^{18}F] 19	[^{18}F] 1	[^{18}F] 19	[^{18}F] 1	[^{18}F] 19
caudate	1.1	1.1	1.4	1.7	1.6	2.5	1.5	2.0
putamen	1.2	1.6	1.7	2.4	1.9	3.5	1.7	2.1
thalamus	1.2	1.5	1.6	2.2	1.8	3.1	1.7	2.4
midbrain	1.4	1.6	1.9	2.5	2.4	4.2	1.6	2.0
pons	0.9	1.2	1.3	2.0	1.6	3.1	1.5	1.6
medulla	1.1	1.5	1.5	2.1	1.9	3.0	1.1	1.5

^a See Tables S6, S7, S12, and S14, Supporting Information, for the complete data. Also see Figures S1, S2, S16, and S18, Supporting Information.

because **18** also has an affinity for the SERT⁶⁰ and we have previously demonstrated that **18** will displace SERT PET tracers.^{63,64}



We have recently reported the microPET and HRRT PET imaging properties of [^{18}F]**19** in nonhuman primates (anesthetized and awake, respectively).⁶⁴ Compound **19** is the para-substituted isomer of **1** and has a higher affinity at all three transporters than **1** (K_i = 0.08 nM SERT, 13 nM DAT, and 28 nM NET) but a lower SERT versus DAT selectivity (~ 162). Comparison between [^{18}F]**19** and [^{18}F]**1** will reveal the effect of placing the vinyl iodide group in the para- or meta-position. The microPET baseline TACs for [^{18}F]**19** are reproduced in Figure S15, Supporting Information, and the awake rhesus monkey HRRT baseline TACs are reproduced in Figure S17, Supporting Information. Table 4 compares the times of peak uptake between [^{18}F]**19** and [^{18}F]**1** for the microPET and HRRT studies with each compound. In both the anesthetized and awake studies [^{18}F]**1** reaches peak uptake significantly faster than [^{18}F]**19** and [^{18}F]**1** also shows a greater washout (also see Tables S1, S2, S11, and S13, Supporting Information). Thus, [^{18}F]**1** has superior imaging kinetics relative to [^{18}F]**19**. Table 5 compares the ratio of uptake in SERT-rich brain regions to

cerebellum uptake for [^{18}F]**1** and [^{18}F]**19** (also see Tables S6, S7, S12, and S14, Supporting Information). Compound [^{18}F]**19** shows higher uptake ratios than [^{18}F]**1** in both anesthetized and awake states, and for both compounds the uptake ratios increase with time throughout the course of the studies (This is represented by Figures S1, S2, S16, and S18, Supporting Information, where the uptake ratio is plotted vs time). The differences in uptake ratio are a result of the differences in kinetics for each tracer. For compound [^{18}F]**1** (Figure 1) washout from the cerebellum begins after ~ 27 min and washout from the SERT-rich brain regions begins after ~ 65 min. This washout then remains fairly steady and the uptake ratios slowly increase until finally stabilizing after ~ 175 min (Figure S1, Supporting Information). In contrast, [^{18}F]**19** reaches peak uptake in the cerebellum after 45 min and then begins to wash out, while uptake in the SERT-rich brain regions continues to increase until 85–105 min followed by a very slow washout. This results in a continuously increasing uptake ratio with time (Figure S16, Supporting Information). Similar differences between [^{18}F]**1** and [^{18}F]**19** are also observed in the awake monkey studies (Figures S2 and S18, Supporting Information). Therefore, [^{18}F]**1** may be more ideally suited for measuring SERT density in humans because of its superior kinetics and the ability to achieve a quasi-equilibrium. Alternatively, as shown in Figure S19, Supporting Information, chasing [^{18}F]**19** with **15** produces a more drastic displacement of [^{18}F]**19** than occurs with [^{18}F]**1** (Figure 9) which suggests that [^{18}F]**19** may be better suited to be used for occupancy determination studies of SSRIs.^{15,25–27} Thus, compounds [^{18}F]**1** and [^{18}F]**19** each have their own unique imaging properties and are both promising candidates for use in human PET studies. The decision of which tracer to use, [^{18}F]**1** or [^{18}F]**19**, would therefore have to be determined by the objective of the pending study.

Summary

The SERT ligands **1–4** were synthesized from *m*-bromophenyl tropane **5** and evaluated for binding to the SERT, DAT, and NET with in vitro competition binding assays using transfected cells. Compounds **1–4** have a high and nearly equal affinity for the SERT and are selective for the SERT over the DAT and NET. Radiolabeling afforded tracers [^{18}F]**1–4** with higher decay-corrected radiochemical yields obtained with [^{18}F]FETOBs than with [^{18}F]FPROBs. Tracers [^{18}F]**1–4** were found to have lipophilicities in the range $\log P_{7.4} = 1.4–1.9$. MicroPET imaging studies in anesthetized cynomolgus monkeys demonstrated that [^{18}F]**1–4** behave very similarly in vivo with peak uptake achieved after 45–55 min, followed by a steady washout. The ratios of uptake in SERT-rich brain regions to cerebellum uptake for [^{18}F]**1–4** were also very similar. An HRRT study was performed with [^{18}F]**1** in an awake rhesus monkey to determine if anesthesia was influencing the behavior of [^{18}F]**1**. The time to peak uptake and the uptake ratio of [^{18}F]**1** were similar between anesthetized and awake states indicating that anesthesia does not affect the imaging properties of [^{18}F]**1**. Chase studies with the SERT ligand **15** displaced tracers [^{18}F]**1–4** but chase studies with the DAT ligand **16** did not displace tracers [^{18}F]**1–4** thus indicating that the observed uptake in the brain is a result of preferential binding to the SERT and not the DAT. Comparison between [^{18}F]**1** and the para-substituted isomer [^{18}F]**19** demonstrates that the position of the vinyl iodide group can have a significant affect on the tracer properties of the two compounds. Peak uptake in the putamen and caudate is achieved after 45 and 55 min, respectively, with [^{18}F]**1** but it takes 95 and 105 min, respec-

tively, with [^{18}F]**19**. Similar effects are observed in the midbrain and thalamus. Alternatively, the ratio of uptake in the SERT-rich brain regions to the cerebellum is higher with [^{18}F]**19** than with [^{18}F]**1**. Both [^{18}F]**1** and [^{18}F]**19** are, therefore, promising candidates for advancement to human studies because [^{18}F]**1** offers faster kinetics than [^{18}F]**19** but [^{18}F]**19** offers higher uptake ratios than [^{18}F]**1** and it will be interesting to see if these differences are maintained in human subjects. Additionally, each compound can be radiolabeled as [^{123}I]**1** or [^{123}I]**19** for SPECT imaging which will further expand the possible uses of these two tracers. We have, therefore, chosen to further pursue [^{18}F]**1** out of this series of tracers reported above in combination with [^{18}F]**19** due to the higher radiochemical yields that can be obtained by *O*-alkylation with [^{18}F]FETOBs, the ability to use both [^{18}F]**1** and [^{18}F]**19** and to make direct comparisons between the two, and the possibility of radiolabeling each with ^{123}I for SPECT imaging. Thus, we intend to evaluate both [^{18}F]**1** and [^{18}F]**19** in healthy normal human volunteers under baseline conditions as well as perform test–retest variability in dose-dependent blocking studies to determine their suitability for measuring occupancy in subjects with neuropsychiatric disorders.

Experimental Section

General. Solvents were purchased from VWR and had originated from EMD or Burdick and Jackson. Anhydrous solvents (100-mL septum-capped bottles) were purchased from Aldrich. TLC plates used were EMD glass-backed Silica Gel 60 F₂₅₄, 20 \times 20 cm, 250 μm . Preparatory TLC plates used were Analtech Uniplat Silica Gel GF 20 \times 20 cm, 2000 μm . Silica gel used was EMD Silica Gel 60, 40–63 μm . Radial chromatography was performed with a Harrison Research Chromatotron. Semipreparatory HPLC: Waters XTerra Prep RP₁₈, 5 μm , 19 \times 100 mm + guard cartridge (19 \times 10 mm), 60:40:0.1 v/v/v MeOH/H₂O/NEt₃. Analytical HPLC: Waters NovaPak 3.9 \times 150 mm, 75:25:0.1 v/v/v MeOH/H₂O/NEt₃. HRMS was performed by the Emory University Mass Spectrometry Center. NMR spectrometry was performed on Varian Inova and Unity spectrometers at the specified frequencies.

3 β -(3'-Bromophenyl)tropane-2 β -carboxylic acid (6). 2 β -Carbomethoxy-3 β -(3'-bromophenyl)tropane (**5**) (0.74 g, 2.19 mmol), 1,4-dioxane (15 mL), and H₂O (15 mL) were stirred at reflux for 16 h. The solvent was removed azeotropically with EtOH to give a white solid that was dried under vacuum for 3 h. The solid was suspended in CHCl₃ (5 mL), cooled to 0 $^{\circ}\text{C}$, and the precipitate was isolated by filtration and dried under vacuum to afford 0.48 g (68%) of a white solid: ^1H NMR (600 MHz, CD₃OD) δ 7.51 (br s, 1 H), 7.36 (d, 1 H, $J = 7.8$ Hz), 7.31 (d, 1 H, $J = 7.8$ Hz), 7.20 (dd, 1 H, $J = 7.8$ Hz), 3.94 (m, 2 H), 3.36 (dt, 1 H, $J = 6.0$ Hz, $J = 13.8$ Hz), 2.81 (m, 1 H), 2.79 (s, 3 H), 2.69 (m, 1 H), 2.40 (m, 2 H), 2.17 (m, 2 H), 1.84 (m, 1 H); HRMS (APCI) [MH]⁺ Calcd for C₁₅H₁₉O₂N⁷⁹Br 324.0594, Found 324.0593; Calcd for C₁₅H₁₉O₂N⁸¹Br 326.0578, Found 326.0574.

2 β -Carbo(2-fluoroethoxy)-3 β -(3'-bromophenyl)tropane (7). 3 β -(3'-Bromophenyl)tropane-2 β -carboxylic acid (**6**) (0.43 g, 1.33 mmol) was suspended in anhydrous CH₂Cl₂ (25 mL) under Ar, followed by addition of anhydrous NEt₃ (0.30 mL, 2.15 mmol, 1.6 equiv.) and cooling to -4 $^{\circ}\text{C}$. Oxalyl chloride (1.0 mL, 2 M CH₂Cl₂, 2.0 mmol, 1.5 equiv.) was diluted with anhydrous CH₂Cl₂ (5 mL) and added dropwise through the condenser to the reaction flask over a period of 5 min. The reaction mixture was stirred under Ar at -4 $^{\circ}\text{C}$ for 20 min, warmed to rt, and the solvent was removed to give a brown solid that was dried under vacuum for 20 min. The solid was dissolved in anhydrous CH₂Cl₂ (25 mL) and cooled to -4 $^{\circ}\text{C}$ under Ar. FETOH (0.80 mL, 13.62 mmol 10.2 equiv.) and anhydrous NEt₃ (0.30 mL, 2.15 mmol, 1.6 equiv.) were dissolved in anhydrous CH₂Cl₂ (10 mL) and added dropwise through the condenser to the acid chloride solution over a period of 3 min. The reaction mixture was stirred under Ar at -4 $^{\circ}\text{C}$ for 5 min, warmed to rt, and stirred at rt for 80 min, and then the solvent was removed to give a brown solid that was dried under

vacuum for 20 min. The solid was dissolved in CH_2Cl_2 , poured onto dry silica (43 mm h \times 43 mm i.d.), and eluted under vacuum: CH_2Cl_2 (100 mL), v/v/v hexane/EtOAc/ NEt_3 75:20:5 (300 mL), 50:45:5 (100 mL) to afford 0.31 g (63%) of a colorless syrup: TLC R_f = 0.38 (50:45:5 v/v/v hexane/EtOAc/ NEt_3); ^1H NMR (600 MHz, CDCl_3) δ 7.39 (s, 1 H), 7.29 (d, 1 H, J = 7.8 Hz), 7.21 (d, 1 H, J = 7.8 Hz), 7.14 (dd, 1 H, J = 7.8 Hz), 4.45 (dddd, 2 H, $^2J_{\text{HF}}$ = 46.8 Hz, $^3J_{\text{HH}}$ = 5.8 Hz, $^3J_{\text{HH}}$ = 2.6 Hz), 4.29 (dddd, 1 H, $^3J_{\text{HF}}$ = 29.4 Hz, $^2J_{\text{HH}}$ = 13.2 Hz, $^3J_{\text{HH}}$ = 5.2 Hz, $^3J_{\text{HH}}$ = 2.6 Hz), 4.08 (dddd, 1 H, $^3J_{\text{HF}}$ = 28.2 Hz, $^2J_{\text{HH}}$ = 12.6 Hz, $^3J_{\text{HH}}$ = 5.9 Hz, $^3J_{\text{HH}}$ = 2.4 Hz), 3.63 (m, 1 H), 3.36 (m, 1 H), 2.98 (m, 2 H), 2.53 (td, 1 H, J = 3.0 Hz, J = 12.3 Hz), 2.22 (s, 3 H), 2.20 (m, 1 H), 2.11 (m, 1 H), 1.70 (m, 2 H), 1.60 (m, 2 H); ^{13}C NMR (150 MHz, CDCl_3) δ 171.29, 145.79, 130.73, 129.69, 129.16, 126.10, 122.38, 81.74 (d, $^1J_{\text{CF}}$ = 169.2 Hz), 65.47, 62.91 (d, $^2J_{\text{CF}}$ = 20.6 Hz), 62.33, 52.65, 42.08, 34.04, 33.71, 25.95, 25.45; HRMS (APCI) $[\text{MH}]^+$ calcd for $\text{C}_{17}\text{H}_{22}\text{O}_2\text{N}^{79}\text{Br}$ 370.0813, found 370.0812; calcd for $\text{C}_{17}\text{H}_{22}\text{O}_2\text{N}^{81}\text{Br}$ 372.0797, found 372.0793.

2 β -Carbo(3-fluoropropoxy)-3 β -(3'-bromophenyl)tropane (8). 3 β -(3'-Bromophenyl)tropane-2 β -carboxylic acid (**6**) (0.11 g, 0.34 mmol) was suspended in anhydrous CH_2Cl_2 (8 mL) under Ar, followed by addition of anhydrous NEt_3 (0.08 mL, 0.57 mmol, 1.7 equiv.) and cooling to -4°C . Oxalyl chloride (0.27 mL, 2 M CH_2Cl_2 , 0.54 mmol, 1.6 equiv.) was diluted with anhydrous CH_2Cl_2 (2 mL) and added dropwise to the reaction flask over a period of 90 s. The reaction mixture was stirred under Ar at -4°C for 20 min, warmed to rt, and the solvent was removed to give a brown solid that was dried under vacuum for 20 min. The solid was dissolved in anhydrous CH_2Cl_2 (10 mL) and cooled to -4°C under Ar. FPrOH (0.32 g, 4.10 mmol, 12.1 equiv.) and anhydrous NEt_3 (0.08 mL, 0.57 mmol, 1.7 equiv.) were dissolved in anhydrous CH_2Cl_2 (5 mL) and added dropwise to the acid chloride solution. The reaction mixture was stirred under Ar at -4°C for 5 min, warmed to rt, and the solvent was removed to give a brown solid that was dried under vacuum for 10 min. The solid was dissolved in CH_2Cl_2 , poured onto dry silica (39 mm h \times 43 mm i.d.), and eluted under vacuum: CH_2Cl_2 (100 mL), v/v/v hexane/EtOAc/ NEt_3 75:20:5 (300 mL), 50:45:5 (100 mL) to afford 88 mg (67%) of a colorless syrup: TLC R_f = 0.41 (50:45:5 v/v/v hexane/EtOAc/ NEt_3); ^1H NMR (600 MHz, CDCl_3) δ 7.36 (s, 1 H), 7.29 (d, 1 H, J = 7.8 Hz), 7.19 (d, 1 H, J = 7.8 Hz), 7.14 (dd, 1 H, J = 7.8 Hz), 4.37 (2 m, 2 H, $^2J_{\text{HF}}$ = 47.1 Hz), 4.14 (m, 1 H), 3.98 (m, 1 H), 3.58 (m, 1 H), 3.35 (m, 1 H), 2.98 (dt, 1 H, J = 5.7 Hz, J = 12.0 Hz), 2.91 (m, 1 H), 2.50 (td, 1 H, J = 2.4 Hz, J = 12.6 Hz), 2.21 (s, 3 H), 2.18 (m, 1 H), 2.11 (m, 1 H), 1.85 (2 m, 2 H, $^3J_{\text{HF}}$ = 24.6 Hz), 1.70 (m, 2 H), 1.60 (m, 1 H); ^{13}C NMR (150 MHz, CDCl_3) δ 171.62, 146.01, 130.57, 129.72, 129.09, 125.98, 122.38, 80.89 (d, $^1J_{\text{CF}}$ = 165.0 Hz), 65.55, 62.29, 59.96 (d, $^3J_{\text{CF}}$ = 6.2 Hz), 52.80, 42.10, 34.11, 33.61, 29.94 (d, $^2J_{\text{CF}}$ = 20.6 Hz), 26.00, 25.43; HRMS (APCI) $[\text{MH}]^+$ Calcd for $\text{C}_{18}\text{H}_{24}\text{O}_2\text{N}^{79}\text{BrF}$ 384.0969, Found 384.0971; Calcd for $\text{C}_{18}\text{H}_{24}\text{O}_2\text{N}^{81}\text{BrF}$ 386.0954, Found 386.0953.

2 β -Carbo(2-fluoroethoxy)-3 β -(3'-bromophenyl)nortropane (9). 2 β -Carbo(2-fluoroethoxy)-3 β -(3'-bromophenyl)tropane (**7**) (0.26 g, 7.02×10^{-4} mol), 2,2,2-trichloroethyl chloroformate (1.0 mL, 7.27 mmol, 10.4 equiv.), Na_2CO_3 (36 mg, 0.34 mmol, 0.5 equiv.), and toluene (15 mL) were stirred at reflux under Ar for 4 h, cooled, poured onto dry silica (43 mm h \times 43 mm i.d.), and eluted under vacuum with CH_2Cl_2 (75 mL) and then 75:20:5 v/v/v hexane/EtOAc/ NEt_3 . The solvent was removed to give a colorless residue that was dried under vacuum (0.37 g). To the residue was added Zn dust (0.46 g), AcOH (10 mL), and H_2O (0.3 mL), and the mixture was stirred at rt for 21 h. The reaction mixture was filtered, the filtrate was diluted with CH_2Cl_2 (50 mL) and H_2O (50 mL), and cooled to 0°C . The aqueous phase was basified to pH 11 with conc. NH_4OH (aq); the layers were separated, and the aqueous layer was extracted with CH_2Cl_2 (20 mL \times 2). The combined CH_2Cl_2 layers were dried over MgSO_4 , and the solvent was removed to give a colorless oil. The oil was dissolved in CH_2Cl_2 , poured onto dry silica (35 mm h \times 43 mm i.d.), and eluted under vacuum: CH_2Cl_2 (75 mL), then v/v/v hexane/EtOAc/ NEt_3 75:20:5 (50 mL), 50:45:5 (50 mL), 20:75:5 (500 mL). The solvent was removed to

give 0.18 g (72%) of a light yellow oil: TLC R_f = 0.10 (50:45:5 v/v/v hexane/EtOAc/ NEt_3); ^1H NMR (600 MHz, CDCl_3) δ 7.35 (s, 1 H), 7.33 (m, 1 H), 7.15 (m, 2 H), 4.36 (dddd, 1 H, $^2J_{\text{HF}}$ = 46.8 Hz, $^2J_{\text{HH}}$ = 10.8 Hz, $^3J_{\text{HH}}$ = 2.4 Hz, $^3J_{\text{HH}}$ = 5.4 Hz), 4.23 (dddd, 1 H, $^2J_{\text{HF}}$ = 48.0 Hz, $^2J_{\text{HH}}$ = 10.8 Hz, $^3J_{\text{HH}}$ = 2.4 Hz, $^3J_{\text{HH}}$ = 6.6 Hz), 4.12 (dddd, 1 H, $^3J_{\text{HF}}$ = 26.4 Hz, $^2J_{\text{HH}}$ = 13.2 Hz, $^3J_{\text{HH}}$ = 2.4 Hz, $^3J_{\text{HH}}$ = 6.6 Hz), 4.01 (dddd, 1 H, $^3J_{\text{HF}}$ = 30.0 Hz, $^2J_{\text{HH}}$ = 13.2 Hz, $^3J_{\text{HH}}$ = 2.4 Hz, $^3J_{\text{HH}}$ = 5.4 Hz), 3.75 (m, 2 H), 3.25 (dt, 1 H, J = 6.0 Hz, J = 12.6 Hz), 2.82 (m, 1 H), 2.37 (td, 1 H, J = 3.0 Hz, J = 13.2 Hz), 2.12 (m, 1 H), 2.01 (m, 1 H), 1.76 (m, 1 H), 1.67 (m, 3 H); ^{13}C NMR (150 MHz, CDCl_3) δ 172.98, 144.85, 130.64, 130.00, 129.82, 126.21, 122.61, 81.25 (d, $^1J_{\text{CF}}$ = 169.2 Hz), 62.99 (d, $^2J_{\text{CF}}$ = 20.7 Hz), 56.54, 53.76, 51.08, 35.44, 33.60, 29.33, 27.89; HRMS (APCI) $[\text{MH}]^+$ Calcd for $\text{C}_{16}\text{H}_{20}\text{O}_2\text{N}^{79}\text{BrF}$ 356.0656, Found 356.0656; Calcd for $\text{C}_{16}\text{H}_{20}\text{O}_2\text{N}^{81}\text{BrF}$ 358.0635, Found 358.0636.

2 β -Carbo(3-fluoropropoxy)-3 β -(3'-bromophenyl)nortropane (10). 2 β -Carbo(3-fluoropropoxy)-3 β -(3'-bromophenyl)tropane (**8**) (78 mg, 2.03×10^{-4} mol), 2,2,2-trichloroethyl chloroformate (0.30 mL, 2.18 mmol, 10.7 equiv.), Na_2CO_3 (11 mg, 0.10 mmol, 0.5 equiv.), and toluene (5 mL) were stirred at reflux under Ar for 4 h, cooled, poured onto dry silica (25 mm h \times 33 mm i.d.), and eluted under vacuum with CH_2Cl_2 (25 mL) and then 75:20:5 v/v/v hexane/EtOAc/ NEt_3 (100 mL). The solvent was removed to give a colorless residue that was dried under vacuum (0.11 g). To the residue was added Zn dust (0.13 g), AcOH (4 mL), and H_2O (0.1 mL), and the mixture was stirred at rt for 21 h. The reaction mixture was filtered; the filtrate was diluted with CH_2Cl_2 (25 mL) and H_2O (25 mL) and cooled to 0°C . The aqueous phase was basified to pH 10 with conc. NH_4OH (aq); the layers were separated, and the aqueous layer was extracted with CH_2Cl_2 (10 mL \times 2). The combined CH_2Cl_2 layers were dried over MgSO_4 and the solvent was removed to give a colorless oil. The oil was dissolved in CH_2Cl_2 , poured onto dry silica (26 mm h \times 33 mm i.d.), and eluted under vacuum: CH_2Cl_2 (75 mL), then v/v/v hexane/EtOAc/ NEt_3 75:20:5 (50 mL), 50:45:5 (75 mL), 20:75:5 (100 mL). The solvent was removed to give 44 mg (59%) of a light yellow oil: ^1H NMR (600 MHz, CDCl_3) δ 7.33 (m, 2 H), 7.15 (m, 2 H), 4.17 (ddd, 2 H, $^2J_{\text{HF}}$ = 47.4 Hz, J_{HH} = 5.7 Hz), 4.01 (m, 1 H), 3.91 (m, 1 H), 3.73 (m, 2 H), 3.24 (dt, 1 H, J = 5.9 Hz, J = 13.2 Hz), 2.73 (d, 1 H, J = 5.4 Hz), 2.36 (td, 1 H, J = 2.4 Hz, J = 13.2 Hz), 2.13 (m, 1 H), 2.02 (m, 1 H), 1.69 (m, 6 H); ^{13}C NMR (150 MHz, CDCl_3) δ 173.30, 144.95, 130.74, 130.04, 129.83, 126.21, 122.61, 80.46 (d, $^1J_{\text{CF}}$ = 165.2 Hz), 60.11 (d, $^3J_{\text{CF}}$ = 5.9 Hz), 56.45, 53.67, 51.00, 35.62, 33.56, 29.68 (d, $^2J_{\text{CF}}$ = 20.7 Hz), 29.22, 27.79; HRMS (APCI) $[\text{MH}]^+$ Calcd for $\text{C}_{17}\text{H}_{22}\text{O}_2\text{N}^{79}\text{BrF}$ 370.0813, Found 370.0814; Calcd for $\text{C}_{17}\text{H}_{22}\text{O}_2\text{N}^{81}\text{BrF}$ 372.0792, Found 372.0793.

(Z)-1,2-Bis(trimethylstannyl)ethane. Purified acetylene (passed successively through a -78°C cold trap, conc. H_2SO_4 (aq), NaOH (s), CaCl_2 (s), and then Drierite) was bubbled through a solution of hexamethylditin (2.52 g, 7.69 mmol), $\text{Pd}(\text{PPh}_3)_4$ (0.89 g, 0.77 mmol), and 1,4-dioxane (20 mL), purged with Ar for 45 min prior to use) at 65°C for 4 h. The solution was cooled to rt, stirred at rt for 20 min, and filtered. The filtrate was poured onto silica gel (14 cm high \times 4 cm i.d.) that had been pretreated with 10% NEt_3 /hexane (100 mL) and then 1% NEt_3 /hexane (100 mL). The product was eluted under vacuum with 1% NEt_3 /hexane (200 mL), and the solvent was removed to give a dark orange oil that was briefly dried under vacuum (2.48 g, 91%). TLC R_f = 0.66 (1% NEt_3 /Hexane); ^1H NMR (400 MHz, CDCl_3) δ 7.33 (s, 2 H), 0.17 (t, 18 H, $^2J_{\text{SnH}}$ = 26.6 Hz); ^{13}C NMR (150 MHz, CDCl_3) δ 155.17, -8.10 .

(Z)-1,2-Bis(trimethylstannyl)ethene was found to be stable (by ^1H NMR spectroscopy) for at least 1 year when stored at -15°C under Ar and protected from moisture and light (the vial was flushed with Ar, capped, wrapped with Parafilm, completely wrapped in Al foil, and then stored in a resealable plastic bag in a freezer).

2 β -Carbo(2-fluoroethoxy)-3 β -(3'-(Z)-2-trimethylstannylethene-nyl)phenyl)nortropane (11). 2 β -Carbo(2-fluoroethoxy)-3 β -(3'-bromophenyl)nortropane (**9**) (0.23 g, 6.46×10^{-4} mol), (Z)-1,2-bis(trimethylstannyl)ethene (0.72 g, 2.04 mmol, 3.2 equiv.), $\text{Pd}(\text{PPh}_3)_4$ (80 mg, 6.92×10^{-5} mol, 0.1 equiv.), and toluene (25

mL, purged with Ar for 2 h) were stirred at reflux under Ar for 18 h. The reaction mixture was cooled to rt, stirred at rt for 1.5 h, poured onto dry silica (40 mm h × 43 mm i.d.), and eluted under vacuum: CH₂Cl₂ (100 mL), hexane (50 mL), v/v/v hexane/EtOAc/NEt₃ 75:20:5 (100 mL), 50:45:5 (100 mL), 20:75:5 (300 mL). The solvent was removed to give an orange oil that was radially chromatographed (4 mm silica, v/v/v hexane/EtOAc/NEt₃ 90:8:2 (300 mL), 75:20:5 (1 L)) to afford 0.13 g of a colorless syrup that was ~73:27 *cis/trans* by integration of the ¹H NMR vinyl resonances. The product was again radially chromatographed (2 mm silica, v/v/v hexane/EtOAc/NEt₃ 95:4:1 (1 L), 90:8:2 (800 mL), 75:20:5 (200 mL), 50:45:5 (200 mL)) to afford 64 mg (21%) of a colorless syrup that was ~93:7 *cis/trans*, 29 mg (10%) of a colorless syrup that was ~76:24 *cis/trans*, and 29 mg (10%) of a colorless syrup that was ~47:53 *cis/trans*. TLC *R_f* = 0.28 (20:75:5 v/v/v hexane/EtOAc/NEt₃); ¹H NMR (600 MHz, CDCl₃) δ 7.54 (td, 1 H, ³J_{HH} = 13.8 Hz, ³J_{SnH} = 75.0 Hz), 7.23 (dd, 1 H, *J* = 7.8 Hz), 7.11 (m, 3 H), 6.18 (td, 1 H, ³J_{HH} = 13.8 Hz, ²J_{SnH} = 32.0 Hz), 4.33 (dddd, 1 H, ²J_{HF} = 46.9 Hz, ²J_{HH} = 10.8 Hz, ³J_{HH} = 5.4 Hz, ³J_{HH} = 2.4 Hz), 4.20 (dddd, 1 H, ²J_{HF} = 47.7 Hz, ²J_{HH} = 10.8 Hz, ³J_{HH} = 6.9 Hz, ³J_{HH} = 2.4 Hz), 4.10 (partially resolved dddd, 1 H, ³J_{HH} = 2.4 Hz), 3.95 (dddd, 1 H, ³J_{HF} = 30.1 Hz, ²J_{HH} = 13.0 Hz, ³J_{HH} = 5.4 Hz, ³J_{HH} = 2.4 Hz), 3.74 (m, 2 H), 3.27 (dt, 1 H, *J* = 5.4 Hz, *J* = 13.2 Hz), 2.83 (m, 1 H), 2.43 (td, 1 H, *J* = 3.0 Hz, *J* = 12.9 Hz), 2.13 (m, 1 H), 2.02 (m, 1 H), 1.77 (m, 1 H), 1.68 (m, 2 H), 1.61 (br s, 1 H), 0.08 (t, 9 H, ²J_{SnH} = 27.3 Hz); ¹³C NMR (100 MHz, CDCl₃) δ 173.18, 147.41, 142.29, 141.13, 133.71, 128.30, 126.86, 126.47, 125.57, 81.22 (d, ¹J_{CF} = 169.1 Hz), 62.85 (d, ²J_{CF} = 20.5 Hz), 56.53, 53.84, 51.24, 35.75, 33.84, 29.37, 27.88, -7.89; HRMS (ESI) [M + H]⁺ Calcd for C₂₁H₃₁O₂NF¹²⁰Sn 468.1355, Found 468.1354.

2β-Carbo(3-fluoropropoxy)-3β-(3'-((Z)-2-trimethylstannylethynyl)phenyl)nortropine (12). 2β-Carbo(3-fluoropropoxy)-3β-(3'-bromophenyl)nortropine (**10**) (0.15 g, 4.05 × 10⁻⁴ mol), (Z)-1,2-bis(trimethylstannyl)ethene (0.54 g, 1.53 mmol, 3.8 equiv), Pd(PPh₃)₄ (73 mg, 6.32 × 10⁻⁵ mol, 0.16 equiv), and toluene (15 mL) were stirred at reflux under Ar for 16 h. The reaction mixture was cooled to rt, poured onto dry silica (40 mm h × 43 mm i.d.) that had been pretreated with 10% NEt₃/hexane (50 mL), and eluted under vacuum: CH₂Cl₂ (50 mL), hexane/EtOAc/NEt₃ v/v/v 75:20:5 (50 mL), 50:45:5 (50 mL), 20:75:5 (400 mL). The solvent was removed to give a brown oil (0.17 g) that was ~78:22 *cis/trans* (+ impurities) by integration of the ¹H NMR vinyl resonances. Purification by radial chromatography (2 mm silica, v/v/v hexane/EtOAc/NEt₃ 95:4:1 (100 mL), 90:8:2 (700 mL), 85:12:3 (100 mL), 80:16:4 (100 mL), and 75:20:5 (300 mL)) afforded 52 mg (27%) of a faint yellow syrup that was ~91:9 *cis/trans* and 41 mg (21%) of a faint yellow syrup that was ~68:32 *cis/trans*. TLC *R_f* = 0.24 (silica, 20:75:5 v/v/v hexane/EtOAc/NEt₃); ¹H NMR (600 MHz, CDCl₃) δ 7.52 (td, 1 H, ³J_{HH} = 13.8 Hz, ³J_{SnH} = 73.4 Hz), 7.22 (dd, 1 H, *J* = 7.5 Hz), 7.10 (s, 1 H), 7.08 (m, 2 H), 6.17 (td, 1 H, ³J_{HH} = 13.8 Hz, ²J_{SnH} = 32.1 Hz), 4.17 and 4.09 (dm, 1 H + 1 H, ²J_{HF} = 46.8 Hz), 3.92 (m, 2 H), 3.73 (m, 1 H), 3.70 (m, 1 H), 3.25 (dt, 1 H, *J* = 5.9 Hz, *J* = 13.2 Hz), 2.73 (m, 1 H), 2.62 (br s, 1 H, NH—conc. dependent), 2.41 (td, 1 H, *J* = 13.2 Hz, *J* = 2.6 Hz), 2.11 (m, 1 H), 2.02 (m, 1 H), 1.76–1.58 (m, 5 H), 0.08 (t, 9 H, ²J_{SnH} = 27.0 Hz); ¹³C NMR (150 MHz, CDCl₃) δ 173.53, 147.32, 142.42, 141.09, 133.71, 128.35, 126.86, 126.50, 125.69, 80.53 (d, ¹J_{CF} = 165.5 Hz), 59.97 (d, ³J_{CF} = 6.2 Hz), 56.52, 53.79, 51.20, 35.97, 33.90, 29.68 (d, ²J_{CF} = 19.9 Hz), 29.33, 27.86, -7.84; HRMS (ESI) [M + H]⁺ Calcd for C₂₂H₃₃O₂NF¹²⁰Sn 482.1512, found 482.1513.

2β-Carbo(2-fluoroethoxy)-3β-(3'-((Z)-2-iodoethenyl)phenyl)nortropine (1). 2β-Carbo(2-fluoroethoxy)-3β-(3'-((Z)-2-trimethylstannylethynyl)phenyl)nortropine (**11**) (~96:4 *cis/trans*, 35 mg, 7.51 × 10⁻⁵ mol) was dissolved in CHCl₃ (5 mL) and cooled to -2 °C under Ar. ICl (0.11 mL, 1.0 M CH₂Cl₂, 0.11 mmol, 1.5 equiv) was added dropwise, the reaction mixture was stirred at -2 °C under Ar for 15 min, and quenched by addition of Na₂S₂O₃·5 H₂O (0.307 g, 1.24 mmol, in 5 mL H₂O). The mixture was diluted with CHCl₃ (5 mL) and H₂O (5 mL); the layers were separated,

and the aqueous phase extracted with CHCl₃ (5 mL × 2). The combined CHCl₃ layers were dried over MgSO₄, the solvent was removed, and the residue was purified by preparative TLC (20:75:5 v/v/v hexane/EtOAc/NEt₃ × 2) to give 23 mg (71%) of a colorless residue that was ~95:5 *cis/trans* by integration of the ¹H NMR vinyl resonances. Separation of the isomers by semipreparative HPLC (cis *t_R* = 19.7 min, 8.8 mL/min) afforded 16 mg (50%) of **1** as a white residue: ¹H NMR (600 MHz, CDCl₃) δ 7.47 and 7.46 (overlapping resonances, 2 H), 7.30 (m, 2 H), 7.20 (d, 1 H, *J* = 7.8 Hz), 6.56 (d, 1 H, *J* = 8.4 Hz), 4.31 (dddd, 1 H, ²J_{HF} = 47.1 Hz, ²J_{HH} = 10.6 Hz, ³J_{HH} = 5.2 Hz, ³J_{HH} = 2.4 Hz), 4.20 (partially resolved ddd, 0.5 H, ²J_{HH} = 10.6 Hz, ³J_{HH} = 7.1 Hz, ³J_{HH} = 2.4 Hz), 4.11 (overlapping resonances, 0.5 H + 0.5 H, ³J_{HH} = 2.4 Hz), 4.06 (ddd, 0.5 H, ²J_{HH} = 12.9 Hz, ³J_{HH} = 7.1 Hz, ³J_{HH} = 2.4 Hz), 3.97 (dddd, 1 H, ³J_{HF} = 30.3 Hz, ²J_{HH} = 12.9 Hz, ³J_{HH} = 5.2 Hz, ³J_{HH} = 2.4 Hz), 3.77 (m, 2 H), 3.32 (dt, 1 H, *J* = 6.0 Hz, *J* = 12.6 Hz), 2.88 (m, 1 H), 2.44 (td, 1 H, *J* = 13.2 Hz, *J* = 3.0 Hz), 2.14 (m, 1 H), 2.03 (m, 1 H), 1.80 (m, 1 H), 1.71 (overlapping m + br s, 2 H + 1 H); ¹³C NMR (150 MHz, CDCl₃) δ 172.90, 141.59, 138.68, 136.93, 128.37, 127.55, 127.45, 126.93, 81.26 (d, ¹J_{CF} = 170.1 Hz), 79.79, 63.26 (d, ²J_{CF} = 18.7 Hz), 56.27, 53.91, 50.76, 35.28, 32.84, 28.71, 27.31; HRMS (ESI) [MH]⁺ calcd for C₁₈H₂₂O₂NF¹²⁷I 430.0674, found 430.0660.

trans-1. ¹H NMR (600 MHz, CDCl₃) δ 7.40 (d, 1 H, *J* = 15.0 Hz), 7.24 (dd, 1 H, *J* = 7.8 Hz), 7.14 (s, 1 H), 7.13 (m, 2 H), 6.82 (d, 1 H, *J* = 15.0 Hz), 4.31 (dddd, 1 H, ²J_{HF} = 47.2 Hz, ²J_{HH} = 10.7 Hz, ³J_{HH} = 5.5 Hz, ³J_{HH} = 2.4 Hz), 4.19 (partially resolved ddd, 0.5 H, ²J_{HH} = 10.7 Hz, ³J_{HH} = 6.7 Hz, ³J_{HH} = 2.4 Hz), 4.11 (overlapping resonances, 0.5 H + 0.5 H, ³J_{HH} = 2.4 Hz), 4.06 (ddd, 0.5 H, ²J_{HH} = 12.9 Hz, ³J_{HH} = 6.7 Hz, ³J_{HH} = 2.4 Hz), 3.95 (dddd, 1 H, ³J_{HF} = 30.1 Hz, ²J_{HH} = 12.9 Hz, ³J_{HH} = 5.5 Hz, ³J_{HH} = 2.4 Hz), 3.77 (m, 2 H), 3.27 (dt, 1 H, *J* = 12.6 Hz, *J* = 6.0 Hz), 2.83 (m, 1 H), 2.62 (br s, 1 H, NH), 2.41 (td, 1 H, *J* = 12.9 Hz, *J* = 3.0 Hz), 2.15 (m, 1 H), 2.04 (m, 1 H), 1.78 (m, 1 H), 1.68 (m, 2 H); semipreparative HPLC (*t_R* = 26.9 min, 8.8 mL/min).

2β-Carbo(2-fluoroethoxy)-3β-(3'-((Z)-2-bromoethenyl)phenyl)nortropine (2). 2β-Carbo(2-fluoroethoxy)-3β-(3'-((Z)-2-trimethylstannylethynyl)phenyl)nortropine (**11**) (~80:20 *cis/trans*, 38 mg, 8.15 × 10⁻⁵ mol) was dissolved in CH₂Cl₂ (4 mL) and cooled to 0 °C under Ar. Br₂ (35 mg, 0.22 mmol) was dissolved in CH₂Cl₂ (1 mL) and added dropwise until a faint yellow color remained. The reaction mixture was stirred at 0 °C under Ar for 6 min and then quenched by addition of Na₂S₂O₃·5H₂O (0.57 g in 4 mL H₂O). The reaction mixture was diluted by addition of CH₂Cl₂ (10 mL) and H₂O (10 mL), the layers were separated, and the aqueous layer was extracted with CH₂Cl₂ (5 mL × 2). The combined CH₂Cl₂ layers were dried over MgSO₄ and the solvent was removed to give a colorless residue that was purified by preparative TLC (silica, 20:75:5 v/v/v hexane/EtOAc/NEt₃) to afford 21 mg (67%) of a colorless residue that was ~91:9 *cis/trans* by integration of the ¹H NMR vinyl resonances. The isomers were separated by semipreparative HPLC (9 mL/min, cis *t_R* = 15.9 min, trans *t_R* = 20.4 min) to afford 16 mg (51%) of **2** as a colorless residue: ¹H NMR (600 MHz, CDCl₃) δ 7.53 (d, 1 H, *J* = 7.8 Hz), 7.49 (s, 1 H), 7.30 (dd, 1 H, *J* = 7.8 Hz), 7.18 (d, 1 H, *J* = 7.2 Hz), 7.04 (d, 1 H, *J* = 8.4 Hz), 6.42 (d, 1 H, *J* = 7.8 Hz), 4.30 (dddd, 1 H, ²J_{HF} = 47.1 Hz, ²J_{HH} = 10.7 Hz, ³J_{HH} = 2.4 Hz, ³J_{HH} = 5.3 Hz), 4.15 (unresolved dddd, 1 H, ²J_{HF} = 47.1 Hz, ²J_{HH} = 2.4 Hz), 4.09 (unresolved dddd, 1 H, ²J_{HH} = 2.4 Hz), 3.96 (dddd, 1 H, ³J_{HF} = 30.6 Hz, ²J_{HH} = 12.9 Hz, ³J_{HH} = 2.4 Hz, ³J_{HH} = 5.3 Hz), 3.82 (br s, 1 H), 3.31 (dt, 1 H, *J* = 6.0 Hz, *J* = 13.2 Hz), 2.87 (m, 1 H), 2.45 (unresolved td, 1 H, *J* = 12.6 Hz), 2.19 (m, 1 H), 2.07 (m, 1 H), 1.80 (m, 1 H), 1.72 (m, 2 H); ¹³C NMR (150 MHz, CDCl₃) δ 173.13, 142.03, 135.13, 132.51, 128.40, 128.10, 127.56, 127.47, 106.71, 81.27 (d, ¹J_{CF} = 169.2 Hz), 63.08 (d, ²J_{CF} = 18.6 Hz), 56.42, 53.89, 51.00, 35.47, 33.31, 29.07, 27.64; Analytical HPLC (*t_R* = 4.4 min, 1 mL/min); HRMS (ESI) [MH]⁺ Calcd for C₁₈H₂₂O₂NF⁷⁹Br 382.0813, Found 382.0815; Calcd for C₁₈H₂₂O₂NF⁸¹Br 384.0792, Found 384.0796.

2β-Carbo(3-fluoropropoxy)-3β-(3'-((Z)-2-iodoethenyl)phenyl)nortropine (3). 2β-Carbo(3-fluoropropoxy)-3β-(3'-((Z)-2-trimethylstannylethynyl)phenyl)nortropine (**12**) (~95:5 *cis/trans*, 84 mg,

1.75×10^{-4} mol) was dissolved in CHCl_3 (10 mL) and cooled to -7°C under $\text{Ar}_{(\text{g})}$. ICI (0.27 mL, 1.0 M CH_2Cl_2 , 0.27 mmol, 1.5 equiv.) was added dropwise, the reaction mixture was warmed to rt, stirred at rt for 10 min, and quenched by addition of $\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$ (0.683 g, 2.75 mmol, in 10 mL H_2O). The reaction mixture was diluted with CHCl_3 (25 mL) and H_2O (25 mL), the layers were separated, and the aqueous layer was extracted with CHCl_3 (10 mL \times 2). The combined CHCl_3 layers were dried over MgSO_4 , and the solvent was removed to give a yellow residue (85 mg) that was purified on Waters silica Sep-Pak Classics (2 in series): loaded with CH_2Cl_2 (1 mL), eluted with CH_2Cl_2 (1 mL), then hexane/EtOAc/ NEt_3 v/v/v 90:8:2 (2 mL), 75:20:5 (5 mL), 50:45:5 (10 mL), 20:75:5 (25 mL). The desired fractions were combined, and the solvent was removed to give 43 mg (55%) of a yellow oil that was \sim 86:14 cis/trans by integration of the ^1H NMR vinyl resonances. The isomers were separated by semipreparative HPLC (8.8 mL/min; cis t_R = 25.2 min; trans t_R = 34.3 min) to afford 26 mg (34%) of **3** as an opaque residue: ^1H NMR (600 MHz, CDCl_3) δ 7.47 and 7.46 (overlapping resonances, 2 H, J = 8.4 Hz), 7.29 (m, 2 H), 7.18 (d, 1 H, J = 7.8 Hz), 6.56 (d, 1 H, J = 8.4 Hz), 4.14 (m, 1 H), 4.06 (m, 1 H), 3.98 (quintet, 1 H, J = 6.0 Hz), 3.87 (m, 1 H), 3.76 (m, 1 H), 3.73 (m, 1 H), 3.31 (dt, 1 H, J = 13.2 Hz, J = 5.4 Hz), 2.78 (d, 1 H, J = 5.4 Hz), 2.53 (br s, 1 H, NH), 2.43 (td, 1 H, J = 2.4 Hz, J = 12.9 Hz), 2.14 (m, 1 H), 2.03 (m, 1 H), 1.78 (m, 1 H), 1.68 (m, 3 H), 1.58 (m, 1 H); ^{13}C NMR (150 MHz, CDCl_3) δ 173.52, 142.47, 138.64, 136.79, 128.32, 127.63, 126.78, 80.55 (d, $^1J_{\text{CF}}$ = 165.0 Hz), 79.56, 59.97 (d, $^3J_{\text{CF}}$ = 4.5 Hz), 56.47, 53.75, 51.19, 35.73, 33.58, 29.70 (d, $^2J_{\text{CF}}$ = 20.6 Hz), 29.31, 27.83; HRMS (ESI) $[\text{MH}]^+$ Calcd for $\text{C}_{19}\text{H}_{24}\text{O}_2\text{NF}^{127}\text{I}$ 444.0830, found 444.0825.

2 β -Carbo(3-fluoropropoxy)-3 β -(3'-((Z)-2-bromoethenyl)phenyl)nortropine (4). 2 β -Carbo(3-fluoropropoxy)-3 β -(3'-((Z)-2-trimethylstannylethenyl)phenyl)nortropine (**12**) (\sim 91:9 cis/trans, 52 mg, 1.08×10^{-4} mol) was dissolved in CHCl_3 (5 mL) and cooled to 0°C under Ar . Br_2 (58 mg, 3.63×10^{-4} mol, 3.4 equiv.) was dissolved in CHCl_3 (1 mL) and added dropwise until a faint yellow color persisted (not all of the solution was added). The reaction mixture was stirred at 0°C under Ar for 15 min and then quenched by addition of $\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$ (98 mg, 3.95×10^{-4} mol, dissolved in 10 mL H_2O). The mixture was diluted with CHCl_3 (10 mL) and H_2O (10 mL); the layers were separated, and the aqueous layer was extracted with CHCl_3 (10 mL \times 2). The combined CHCl_3 layers were dried over MgSO_4 , and the solvent was removed to give a colorless syrup that became an opaque residue when dried under vacuum. The residue was dissolved in CH_2Cl_2 , poured onto dry silica (26 mm h \times 33 mm i.d.), and eluted under vacuum: CH_2Cl_2 (25 mL), hexanes/EtOAc/ NEt_3 v/v/v 75:20:5 (25 mL), 50:45:5 (50 mL), 20:75:5 (250 mL). The isolated product was further purified by preparative TLC (20:75:5 v/v/v hexanes/EtOAc/ NEt_3) to give a faint yellow syrup (26 mg) that was \sim 93:7 cis/trans by integration of the ^1H NMR vinyl resonances. The isomers were separated by semipreparative HPLC (cis t_R = 20.3 min, 9.0 mL/min) to afford 21 mg (49%) of a colorless residue: ^1H NMR (600 MHz, CDCl_3) δ 7.53 (d, 1 H, J = 7.2 Hz), 7.49 (s, 1 H), 7.30 (dd, 1 H, J = 7.2 Hz), 7.15 (d, 1 H, J = 7.2 Hz), 7.04 (d, 1 H, J = 8.4 Hz), 6.44 (d, 1 H, J = 8.4 Hz), 4.11 (m, 1 H), 4.03 (m, 1 H), 3.97 (quintet, 1 H, J = 5.9 Hz), 3.87 (m, 2 H), 3.82 (m, 1 H), 3.32 (dt, 1 H, J = 6.0 Hz, J = 12.6 Hz), 2.80 (overlapping resonances: br s + m, 2 H), 2.46 (td, 1 H, J = 2.4 Hz, J = 13.2 Hz), 2.25 (m, 1 H), 2.14 (m, 1 H), 1.82 (m, 1 H), 1.74 (m, 2 H), 1.65 (m, 1 H), 1.54 (m, 1 H); ^{13}C NMR (150 MHz, CDCl_3) δ 173.55, 141.61, 135.24, 132.32, 128.51, 128.11, 127.69, 127.51, 106.90, 80.41 (d, $^1J_{\text{CF}}$ = 166.0 Hz), 60.38 (d, $^3J_{\text{CF}}$ = 4.2 Hz), 56.31, 53.86, 50.64, 35.63, 33.00, 29.57 (d, $^2J_{\text{CF}}$ = 18.7 Hz), 28.63, 27.34; HRMS (ESI) $[\text{M} + \text{H}]^+$ Calcd for $\text{C}_{19}\text{H}_{24}\text{O}_2\text{FN}^{79}\text{Br}$ 396.0969, found 396.0974.

[^{18}F]Fluoroethylbrosylate. H^{18}F was produced with a Siemens 11-MeV RDS 112 cyclotron by employing the $^{18}\text{O}(\text{p,n})^{18}\text{F}$ reaction in H_2^{18}O . The $\text{H}^{18}\text{F}_{(\text{aq})}$ was transferred to a chemical processing control unit (CPCU), collected on a trap/release cartridge, released with $\text{K}_2\text{CO}_3_{(\text{aq})}$ (0.9 mg in 0.6 mL H_2O), and added to a CH_3CN solution of Kryptofix 222 (5 mg in 1 mL). The solution was placed

in a 110°C oil bath, the solvent was evaporated under a $\text{N}_{2(\text{g})}$ flow, and CH_3CN (3 mL) was added and evaporated in order to azeotropically dry the Kryptofix 222/ K^{18}F . 1,2-Dibrosylethane (4 mg in 1 mL CH_3CN) was added, the reaction mixture was heated at 90°C for 10 min, and the [^{18}F]fluoroethylbrosylate was trapped on a Waters silica Sep-Pak Classic (WAT051900) (previously prepped with 10 mL EtOEt). The [^{18}F]fluoroethylbrosylate was eluted with EtOEt, the EtOEt solution was transferred to a hot cell under $\text{N}_{2(\text{g})}$ pressure and collected in a V-tube to give [^{18}F]fluoroethylbrosylate in 74% radiochemical yield (decay corrected from transfer of $\text{H}^{18}\text{F}_{(\text{aq})}$ to the CPCU). The V-tube was placed in an 80°C oil bath and the EtOEt was evaporated with an $\text{Ar}_{(\text{g})}$ flow. The solution of radiolabeling precursor was then added to this V-tube (see below).

[^{18}F]Fluoropropylbrosylate. Prepared in an analogous manner as [^{18}F]fluoroethylbrosylate using 1,3-dibrosylpropane.

2 β -Carbo(2-[^{18}F]fluoroethoxy)-3 β -(3'-((Z)-2-iodoethenyl)phenyl)nortropine ([^{18}F]1). *N*-(*t*-butoxycarbonyl)-3 β -(3'-((Z)-2-iodoethenyl)phenyl)nortropine-2 β -carboxylic acid (**13**) (\sim 1.1 mg, \sim 98:2 cis/trans) was dissolved in DMF (0.3 mL), deprotonated by addition of 0.1 M $\text{Bu}_4\text{NOH}_{(\text{aq})}$ (16 μL , 0.7 equiv), and added to [^{18}F]fluoroethylbrosylate. The solution was heated at 90°C for 10 min, 6 M $\text{HCl}_{(\text{aq})}$ (\sim 0.16 mL, \sim 422 equiv) was added, the solution was heated at 90°C for 10 min, cooled to 0°C , and neutralized by addition of 6 M $\text{NH}_4\text{OH}_{(\text{aq})}$ (\sim 0.16 mL, \sim 422 equiv). The solution was diluted with HPLC solvent and purified by semipreparative HPLC (t_R (range) = 15–19 min, 9.2 mL/min). The desired fractions were combined, diluted 1:2 v/v with H_2O and loaded onto a Waters C_{18} Sep-Pak. The Sep-Pak was washed with 0.9% $\text{NaCl}_{(\text{aq})}$ (40 mL) and then EtOH (0.5 mL). The product was eluted from the Sep-Pak with EtOH (1.5 mL) and collected in a sealed sterile vial containing 0.9% $\text{NaCl}_{(\text{aq})}$ (3.5 mL). This solution was passed successively through a 1 μm filter and then a 0.2 μm filter (Acrodisc PTFE) under Ar -pressure and collected in a sealed sterile dose vial containing 0.9% $\text{NaCl}_{(\text{aq})}$ (10 mL). The total synthesis time was \sim 80 min from the delivery of [^{18}F]fluoroethylbrosylate to the hot cell with a $6.3 \pm 1.8\%$ (n = 4) radiochemical yield (decay corrected). The product was then analyzed by analytical HPLC (t_R = 4.9 min, 1 mL/min) to determine the radiochemical purity ($97 \pm 2\%$, n = 4).

2 β -Carbo(2-[^{18}F]fluoroethoxy)-3 β -(3'-((Z)-2-bromoethenyl)phenyl)nortropine ([^{18}F]2). *N*-(*t*-butoxycarbonyl)-3 β -(3'-((Z)-2-bromoethenyl)phenyl)nortropine-2 β -carboxylic acid (**14**) (\sim 0.6 mg, \sim 95:5 cis/trans) was dissolved in DMF (0.3 mL), deprotonated by addition of 0.1 M $\text{Bu}_4\text{NOH}_{(\text{aq})}$ (10 μL , 0.7 equiv), and added to [^{18}F]fluoroethylbrosylate. The solution was heated at 90°C for 10 min, 6 M $\text{HCl}_{(\text{aq})}$ (\sim 0.11 mL, \sim 480 equiv) was added, the solution was heated at 90°C for 10 min, cooled to 0°C , and neutralized by addition of 6 M $\text{NH}_4\text{OH}_{(\text{aq})}$ (\sim 0.11 mL, \sim 480 equiv). The solution was diluted with HPLC solvent and purified by semipreparative HPLC (t_R (range) = 10–16 min, 9.2 mL/min). The desired fractions were combined, diluted 1:2 v/v with H_2O and loaded onto a Waters C_{18} Sep-Pak. The Sep-Pak was washed with 0.9% $\text{NaCl}_{(\text{aq})}$ (40 mL) and then EtOH (0.5 mL). The product was eluted from the Sep-Pak with EtOH (1.5 mL) and collected in a sealed sterile vial containing 0.9% $\text{NaCl}_{(\text{aq})}$ (3.5 mL). This solution was passed successively through a 1 μm filter and then a 0.2 μm filter (Acrodisc PTFE) under Ar -pressure and collected in a sealed sterile dose vial containing 0.9% $\text{NaCl}_{(\text{aq})}$ (10 mL). The total synthesis time was \sim 75 min from the delivery of [^{18}F]fluoroethylbrosylate to the hot cell with a $4.2 \pm 2.7\%$ (n = 3) radiochemical yield (decay corrected). The product was then analyzed by analytical HPLC (t_R = 4.4 min, 1 mL/min) to determine the radiochemical purity ($95 \pm 2\%$, n = 3).

2 β -Carbo(3-[^{18}F]fluoropropoxy)-3 β -(3'-((Z)-2-iodoethenyl)phenyl)nortropine ([^{18}F]3). *N*-(*t*-butoxycarbonyl)-3 β -(3'-((Z)-2-iodoethenyl)phenyl)nortropine-2 β -carboxylic acid (**13**) (\sim 0.6 mg) was dissolved in DMF (0.3 mL), deprotonated by addition of 0.1 M $\text{Bu}_4\text{NOH}_{(\text{aq})}$ (11 μL , 0.9 equiv), and added to [^{18}F]fluoropropylbrosylate. The solution was heated at 105°C for 10 min, and 6 M $\text{HCl}_{(\text{aq})}$ (\sim 0.12 mL, \sim 580 equiv) was added. The solution was

heated at 105 °C for 10 min, cooled to 0 °C, and neutralized by addition of 6 M $\text{NH}_4\text{OH}_{(\text{aq})}$ (~0.12 mL, ~580 equiv). The solution was diluted with HPLC solvent and purified by semipreparative HPLC (t_R (range) = 21–25 min, 9.2 mL/min). The desired fractions were combined, diluted 1:2 v/v with H_2O , and loaded onto a Waters C_{18} Sep-Pak. The Sep-Pak was washed with 0.9% $\text{NaCl}_{(\text{aq})}$ (40 mL) and then EtOH (0.5 mL). The product was eluted from the Sep-Pak with EtOH (1.5 mL) and collected in a sealed sterile vial containing 0.9% $\text{NaCl}_{(\text{aq})}$ (3.5 mL). This solution was passed successively through a 1 μm filter and then a 0.2 μm filter (Acrodisc PTFE) under Ar-pressure and collected in a sealed sterile dose vial containing 0.9% $\text{NaCl}_{(\text{aq})}$ (10 mL).

The total synthesis time was ~80 min from the delivery of [^{18}F]fluoropropylbrosylate to the hot cell with a $1.9 \pm 0.8\%$ ($n = 3$) radiochemical yield (decay corrected). The product was then analyzed by analytical HPLC ($t_R = 6.3$ min, 1 mL/min) to determine the radiochemical purity ($94 \pm 4\%$, $n = 3$).

2 β -Carbo(3-[^{18}F]fluoropropoxy)-3 β -(3'-(Z)-2-bromoethenyl)phenyl)nortropine (**[^{18}F]4**). *N*-(*t*-Butoxycarbonyl)-3 β -(3'-(Z)-2-bromoethenyl)phenyl)nortropine-2 β -carboxylic acid (**14**) (~0.6 mg) was dissolved in DMF (0.3 mL), deprotonated by addition of 0.1 M $\text{Bu}_4\text{NOH}_{(\text{aq})}$ (12 μL , 0.9 equiv), and added to [^{18}F]fluoropropylbrosylate. The solution was heated at 90 °C for 10 min, 6 M $\text{HCl}_{(\text{aq})}$ (~0.15 mL, ~655 equiv) was added, the solution was heated at 90 °C for 10 min, cooled to 0 °C, and neutralized by addition of 6 M $\text{NH}_4\text{OH}_{(\text{aq})}$ (~0.15 mL, ~655 equiv). The solution was diluted with HPLC solvent and purified by semipreparative HPLC (t_R (range) = 15–20 min, 9.2 mL/min). The desired fractions were combined, diluted 1:2 v/v with H_2O and loaded onto a Waters C_{18} Sep-Pak. The Sep-Pak was washed with 0.9% $\text{NaCl}_{(\text{aq})}$ (40 mL) and then EtOH (0.5 mL). The product was eluted from the Sep-Pak with EtOH (1.5 mL) and collected in a sealed sterile vial containing 0.9% $\text{NaCl}_{(\text{aq})}$ (3.5 mL). This solution was passed successively through a 1 μm filter and then a 0.2 μm filter (Acrodisc PTFE) under Ar-pressure and collected in a sealed sterile dose vial containing 0.9% $\text{NaCl}_{(\text{aq})}$ (10 mL). The total synthesis time was ~77 min from the delivery of [^{18}F]fluoropropylbrosylate to the hot cell with a $1.7 \pm 0.3\%$ ($n = 3$) radiochemical yield (decay corrected). The product was then analyzed by analytical HPLC ($t_R = 6.0$ min, 1 mL/min) to determine the radiochemical purity ($93 \pm 7\%$, $n = 3$).

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Supporting Information Available: Additional microPET data and elemental analysis data of **1–4**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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